



COURSE DESCRIPTION CARD - SYLLABUS

Course name

Instrumental Analysis - Laboratory - The use of UV-VIS spectrophotometric and electroanalytical techniques in the pharmaceutical analysis

Course

Field of study	Year/Semester
Pharmaceutical Engineering	2/4
Area of study (specialization)	Profile of study
-	general academic
Level of study	Course offered in
First-cycle studies	polish
Form of study	Requirements
full-time	elective

Number of hours

Lecture	Laboratory classes	Other (e.g. online)
0	30	0
Tutorials	Projects/seminars	
0	0	

Number of credit points

2

Lecturers

Responsible for the course/lecturer:

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Responsible for the course/lecturer:

Wydział Technologii Chemicznej

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Prerequisites

Good knowledge of inorganic, analytical and instrumental chemistry, apparatus used in the chemical laboratory, mathematical tools used in the chemical calculations.

Usage a of chemical apparatus and volumetric glassware.

Course objective

The use of instrumental techniques (more detailed than in basic course): UV-VIS spectrophotometry, electroanalytical tchniques,qualitative and quantitative analysis. Assessment of the results of chemical analysis.

Course-related learning outcomes

Knowledge

1. Student has good knowledge in the field of chemistry for the understanding of phenomena and processes occurring during analysis, K_W4
2. Student has theoretically founded good knowledge in the field of analytical chemistry and instrumental analysis K_W04
3. Knows classical and instrumental methods used in assessing the quality of substances for pharmaceutical purposes and in quantitative analysis in medicinal products K_W7

Skills

1. Student can obtain therelevant information from the literature to conduct the determination of an analyte in the real sample. K_U01
2. Student is able to perform chemical analysis, interprets the results of analyzes and draw appropriate conclusions K_U2, K_U3, K_U5, K_U10

Social competences

1. Students can understand the need for self-education and raising their competences in the field of instrumental analysis, K_K1
2. Student is able to work both individually and in team during the laboratory work, K_K2

Methods for verifying learning outcomes and assessment criteria

Learning outcomes presented above are verified as follows:

A series of laboratory exercises of instrumental analysis is preceded by checking the theoretical foundations of the methods used (carried out in a stationary or remote mode via e-Kursy platform). Students prepare written reports on completed exercises.

Programme content

Practical application of spectrophotometric and electroanalytical techniques to determine selected analytes in pharmaceutical and medical samples. Signal measurement methods, analytical



characteristics of the method, application of the method. Chemical calculations necessary in laboratory practice.

Teaching methods

Laboratory classes: analyte determinations using analytical apparatus in accordance with the instructor's instructions.

Bibliography

Basic

1. D.A. Skoog, D.M. West, F.J. Holler, S.R. Crouch, Podstawy Chemii Analitycznej T. 1 i 2, PWN, Warszawa, (1) 2006, (2) 2007
2. J. Minczewski, Z. Marczenko, Chemia Analityczna. Analiza Instrumentalna T. 1-3, PWN, Warszawa, 1,2 (2007), 1(1985)
3. A. Cygański, Chemiczne metody analizy ilościowej, WNT Warszawa, 2019
4. A. Cygański, Metody elektroanalityczne, WNT, Warszawa, 1999
5. I. Baranowska (red.) Analiza śladowa – Zastosowania, Wydawnictwo MALAMUT, Warszawa, 2013
6. J. Namieśnik, P. Konieczka, B. Zygmunt, Ocena i kontrola jakości wyników analitycznych, WNT, 2014.
7. A. Cygański, B. Ptaszyński, J. Krystek, Obliczenia w chemii analitycznej, WNT Warszawa, 2004
8. M. Wesołowski, K. Szefer, D. Zimna, Zbiór zadań z analizy chemicznej, WNT Warszawa, 2002

Additional

1. W. Ufnalski, Równowagi jonowe, WNT Warszawa 2004
2. A. Hulanicki, Reakcje kwasów i zasad w chemii analitycznej, WN PWN Warszawa 2012
3. Z. Galus, Ćwiczenia rachunkowe z chemii analitycznej, WN PWN Warszawa 2020
4. J. Dojlido, J. Zerbe, Instrumentalne metody badania wody i ścieków, Arkady, Warszawa 1997

Breakdown of average student's workload

	Hours	ECTS
Total workload	60	2,0
Classes requiring direct contact with the teacher	40	1,3
Student's own work (literature studies, preparation for laboratory classes, preparation for test) ¹	20	0,7

¹ delete or add other activities as appropriate